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ABSTRACT

Evaluation of changes in the coupled chemical and mechanical properties of reservoir and seal rocks is critical for ensuring both short and long-term security of injected CO_2 . Natural analogs are advantageous for understanding these properties on longer time scales than possible using laboratory or numerical experiments. The Crystal Geyser site in Utah was studied to assess the potential for leakage via fracturing or capillary failure of reservoir and seal rocks altered by natural, long-term CO_a-water-rock interactions.

Fracture mechanics testing using the double torsion method was performed on a suite of naturally altered and unaltered rocks exposed at Crystal Geyser. Fracture toughness measurements demonstrate that CO₂-related alteration has weakened one reservoir sandstone unit by approximately 50%, but did not affect subcritical index. In contrast, the fracture toughness measured of a weak, poorly-cemented sandstone unit was increased due to enhanced calcite cementation, and a similar effect was observed in shale samples tested.

Mercury intrusion capillary pressure (MICP) analyses on shale from a fault-perpendicular transect show relatively low capillary seal capacity nearest the fault where CO₃-alteration is most intense. Šeal capacity increases by nearly an order of magnitude where measured bulk calcite is highest then gradually declines laterally away from the fault. SEM imaging shows matrix replacement with calcite is primarily responsible for increased seal capacity.

This study demonstrates that CO₂-water-rock interaction driven by changes in the geochemical environment have significantly changed rock geomechanical and flow properties over geologic time scales. Alteration and dissolution of grains and cements in reservoir and seal rock can potentially lead to leakage by favoring fracturing and/or lowering capillary seal capacity. Alternatively, where CO, -rich fluids drive significant precipitation of carbonate minerals, fracture growth may be hindered and capillary seal capacity increased.

> RIGHT: Block diagram of Crystal Geyser field site, Utah. Geology constrained by mapping (Urquhart, 2011), published data, and detailed study of the Moab fault found 50 km to the SE (Eichhubl et al., 2009).

III- MERCURY INTRUSION CAPILLARY PRESSURE ANALYSIS (MICP) Effect of CO₂-alteration on seal capacity

The flow properties, permeability, and seal capacity of seal rocks such as shales may change due to CO₂-water-rock interactions. Coupled chemical-mechanical effects have implications for long term storage and injection scenarios. Decreases in seal capacity over time due to alteration may lead to leakage. Conversely, alteration that leads to precipitation of new ninerals may increase seal capacity, enhancing storage security.

MICP analysis is a standard approach for measuring capillary entry pressure (Pe), permeability (k), grain density (ρ), and pore aperture size distributions of fine-grained rocks such as shale. *EQ*: $Pe = (2 \gamma \cos r)/\theta$

where Pe is capillary entry pressure (in Pa), y is surface tension (in N/m), θ is wetting angle (in degrees), and r is average pore throat radius (in m)

Samples were taken from a fault-perpendicular transect in the Mancos Shale and bulk composition measured using XRD.









1. Bureau of Economic Geology, Jackson School of Geosciences, The University of Texas at Austin, TX, United States 2. Geomechanics, Sandia National Laboratories, Albuquerque, NM, United States

1-INTRODUCTION

BIG QUESTION:

1- How does CO₂ alter the geomechanical properties of rock over long time scales, potentially leading to leakage of subsurface reservoirs?

- Two failure mechanisms:
- a. Fracturing

b. Capillary failure

APPROACH:

1- Characterize a range of rocks naturally altered by CO₂-water-rock interactions over geologic time scales: XRD, stable isotopes, petrography

2- Measure and compare fracture mechanics parameters: fracture toughness (K_{IC}),

subcritical index (SCI)

3- Measure and compare seal/flow properties: mercury-intrusion capillary pressure analysis (MICP)





After Hintze & Kowallis, 2009

ABOVE: Stratigraphy of the Green River. Utah area with main units of study highlighted.

(BELOW) confirms that mineral dissolution and precipitation has altered pore networks in the shale, changing its flow properties.



Capillary entry pressures (α to seal capacity) and bulk calcite content plotted against distance (alteration intensity)

150 m from fault: CG 5-29-13 #4 (unaltered Mancos Shale)

Fracture mechanics and seal capacity in a natural CO, system

Jonathan R Major¹, Peter Eichhubl¹, Thomas A. Dewers²



II- FRACTURE MECHANICS TESTING Geomechanical effect of CO₂-related alteration

Fracture mechanics parameters, the fracture toughness (K_{IC}) and subcritical index (SCI) of reservoir and seal units are important inputs for assessing storage capacity, leakage potential, and planned injection rates for numerical models of carbon sequestration sites.

We evaluate whether CO₂-related alteration affect these parameters by testing naturally altered rocks. These rocks may better represent the end-products of alteration than what can be measured under laboratory conditions and time scales.

Fracture toughness describes the ability of a material containing a crack to resist fracture, or in other words, the fracture toughness is the stress intensity required to initiate new fracturing.

Subcritical crack growth may be an important mechanism for and control on fracturing in reservoir and seal units at Crystal Geyser and in subsurface CO, reservoirs. Subcritical crack growth is largely controlled by stress corrosion, which is related to chemical reactions at the fracture tip. The subcritical crack index, n, is the exponent in the power law relationship below and describes the velocity dependence of fractures propagating under subcritical conditions:

 $V = A\left(\frac{K_I}{K_{IC}}\right)$

where

- V : fracture propagation velocity
- : mode-I stress intensity factor,
- \mathbf{K}_{ic} : mode-I critical stress intensity factor
- (fracture toughness)
- A : a pre-exponential constant (Atkinson, 1984; Swanson 1984)

 K_{ic} = fracture toughness Together, these parameters have been shown to control the geometry of emerging fracture patterns, such as fracture lengths, spacing, and clustering, which in turn may effect connectivity and flow properties (Olson, 1993 & 2004).





IV-IDENTIFY AND MEASURE EXTENT OF ALTERATION Stable isotope data

Region 1 Region 2 Region 3

Log Stress Intensity Factor

K^{*}, = stress corrosion limit

crack

crack

XRD data strongly suggests that the alteration at Crystal Geyser is related to interaction with CO_g-rich fluids, but weathering reactions can also effect rock mineralogy in similar ways. Stable isotope data can help confirm that the rocks at Crystal Geyser are effected by CO₂-water-rock interactions.



Stable isotope data of shale samples taken from a fault-perpendicular transect at the Crystal Geyser field site. Elevated δ 13C values are indicative of CO₂-related alteration. Background samples plot near 0‰ (PDB), as expected for a marine shale. Scatter is greater in δ18O values, but shows similar trends. Light isotopes preferentially fractionate with degassing CO₂, and heavier isotopes tend to remain in calcite and other precipitated minerals.

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V- CONCLUSIONS AND IMPLICATIONS

Impact of CO₂ -related alteration on reservoir & seal rocks:

- Geomechanical effects:
- o CO₂ alteration which is characterized by mineral dissolution and precipitation of secondary clays can weaken rocks to fracturing by up to 50% in sandstone
- o CO₂ alteration which is characterized by precipitation of carbonate cement
- can increases K_{IC} in shale samples by ~30-50%
- Permeability & seal capacity :
- o Intense alteration lowers seal capacity of shale by nearly half an order of magnitude through dissolution of grains and cement
- o Enhanced precipitation of carbonate cements in shale increases seal capacity by 3-5X
- o CO₂ alteration significantly impacts permeability in all lithologies

IMPLICATIONS

- Subsurface CO₂ injection and storage involves inherently coupled chemical-mechanical processes
- Diagenetic reactions affect fracture mechanics properties and matrix flow of reservoir and seal rocks
- 6. Comparable chemical reactions can also affect reservoir and seal rocks in EOR operations, wastewater and CO_a injection because fluids are out of chemical equilibrium with the host formation

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